

## Antimicrobial effect of Silver Nanoparticles Synthesized by Chemical Reduction on some Bacteria and Fungi

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### Abstract

**Background;** Nanotechnology involves the tailoring of materials at atomic level to attain unique properties, which can be suitably manipulated for the. Most of the natural processes also take place in the nanometer scale regime.

**Aim;** The purpose from this experiment was to describe antimicrobial effects of silver nanoparticles on some bacteria and fungi.

**Method;** The morphology and size of particles of nanoparticles were determined by scanning electron microscopy (SEM) and Visible absorption spectrophotometer.

**Results;** The antimicrobial effect was studied using the well agar diffusion and broth dilution techniques. The MIC of silver nanoparticles were 30 ppm for *Candida albicans*, *Candida tropicalis* and *Candida Krusei*. Compared to other strains (*E. coli*, *Pseudomonas aureginosa*, *S. aureus*) 100 ppm. The antimicrobial activity was observed against all tested microorganisms at different concentration of nano silver. The Ag NPs causes growth delay for treated isolated with different concentration used in the experiment (increased concentration, increased growth inhibition).

**Conclusion;** The study revealed that silver nanoparticles can be further developed as an antimicrobial agent, hence high activity against microorganisms.

**Keyword;** Antimicrobial effect, Silver Nanoparticles, bacteria and fungi

### INTRODUCTION

Nanotechnology involves the tailoring of materials at atomic level to attain unique properties, which can be suitably manipulated for the (1-3) Most of the natural processes also take place in the nanometer scale regime. Therefore, a confluence of nanotechnology and biology can address several biomedical problems, and can revolutionize the field of health and medicine (4-5). The Ag-NPs have distinctive physical and chemical properties, such as, chemical stability, high thermal and electrical conductivity, nonlinear optical behavior, surface

- enhanced Raman scattering and catalytic activity (6-7). These properties made this nanoparticles to the top of the priority list, to be used in inks, in electronics, and of course biology(8-12). For a long time silver ion has been used as antimicrobial agent since time immemorial in the form of metallic silver nitrate, silver sulfadiazine for the treatment of wounds, burns due to its strong inhibiting effect on bacteria, Viruses, fungi, protozoa(13-22). Due to their proven antimicrobial properties, Ag-NPs are widely used in the daily used commercial products, such as cosmetics, detergent, plastics, food packaging, soaps, pastes, food, and textiles, which has increased their market value to a great extent (23-25) **Aim;** The purpose from this experiment was to describe antimicrobial effects of silver nanoparticles on some bacteria and fungi.

### MATERIALS

#### Microorganisms

In this study three isolates of bacteria; *E. coli*, *S. aureus* (MRSA), *Pseudomonas aeruginosa* and three fungal strains, *Candida albicans*, *Candida tropicalis*, and *Candida krusei*. The bacteria were grown in the Nutrient broth at 37°C, using Nutrient agar slants for sub culturing and Sabourauds broth at 37 °C for fungi using Sabourauds agar slants for subculturing, and preserved all the isolates in refrigerator at 4°C.

#### Synthesis of silver nanoparticles:

The AgNPs prepared from a volume of 100 ml silver nitrate solution with  $1 \times 10^{-4}$  mol / l was mixed with 300 ml of  $2 \times 10^{-2}$  mol / l sodium borohydride both solution was immersed in ice bath. The reaction protected from light. After the addition of silver nitrate solution was finished, the mixture was then vigorously stirred during 15min.

#### Characterization of silver nanoparticles

Silver nanoparticles were characterized spectrophotometrically using vis spectroscopy analyses. Scanning electron microscopy (SEM) has been employed to

characterize the shape and morphologies of formed synthesized of AgNPs.<sup>(26)</sup>

#### Antimicrobial activity of silver nanoparticles: Determination of Minimum inhibitory concentration and Minimum Bactericidal / fungicidal concentration.

Inoculation suspensions of the *Candida albicans*, *Candida tropicalis*, *Candida krusei* were prepared by picking colonies from 24 h Sabraudous agar at 37°C<sup>0</sup>. Also *S. aureus*, *E. coli*, *P. aeruginosa* were prepared by picking colonies from nutrient agar (24h) and using sterile saline (0.85%) v/v NaCl to determine turbidity equivalent to 0.5 McFarland standard ( $1 \times 10^8$ ) colony forming units /ml.

Broth dilution method was followed for measurement of MIC values. Different concentration of nanoparticles silver solution (10,20,30,40,50,100, 150, 250, 300, 400 ppm) were added to LB broth medium, the test tubes were incubated at 37°C<sup>0</sup>. The MIC values (Bacteriostatic / Fungistatic) were estimated the lowest concentration in the test tube that showed no turbidity (growth of isolate) after incubation. The minimum bactericidal / fungicidal concentration (MBC / MFC) was evaluated through sub culturing from each test tube showing no apparent growth on the agar after further 24hr incubation.

#### Antimicrobial susceptibility testing using Agar well diffusion assay NCCL<sup>(27)</sup>

The antibacterial susceptibility of prepared AgNPs against fungi isolates and bacteria isolates were evaluated by Agar well diffusion method. Loop full of freshly – grown bacterial / fungal isolates were inoculated in LB(Luria broth) incubated at 37°C<sup>0</sup> for 24hr. The bacterial suspensions were diluted with normal saline. A just the turbidity and compare with standard tube (McFarland number 0.5) to yield a uniform suspension containing  $1.5 \times 10^8$  CFU / ml and spread the suspension by cotton swab, streaking on LB agar Media were cut into four wells (5 mm diameter) by cork borer and add 30µ of suspension (different concentrations) (the plates were performed in triplicates). All plates of the tested organisms was then allowed to incubate at 37°C<sup>0</sup> for overnight. Reading the result by the measuring the distance / inhibition zones around the wells.

## RESULTS AND DISCUSSION

### UV- visible analysis

The absorption spectrum was measured immediately after preparation. UV – visible spectroscopy is one of the most widely and sensitive techniques for structural characterization of silver nanoparticles synthesis, Figure (1) shows the maximum absorption peak was observed at about 413nm

which is the characteristic absorption peak for Ag nanoparticles.

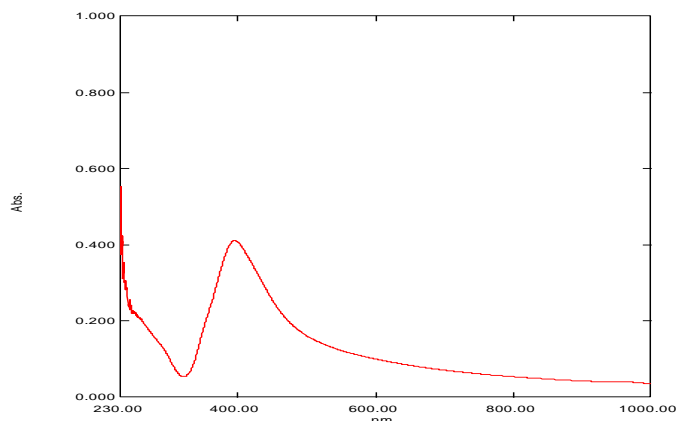


Figure 1: UV – Vis spectra- The maximum absorbance was at 413 nm

### SEM analysis :

SEM was used for observing the morphology of synthesized AgNPs. Fig.(2) shows the SEM images of AgNPs. As, It can be seen that the inner diameter of the AgNPs is ranging between (65-72) nm.

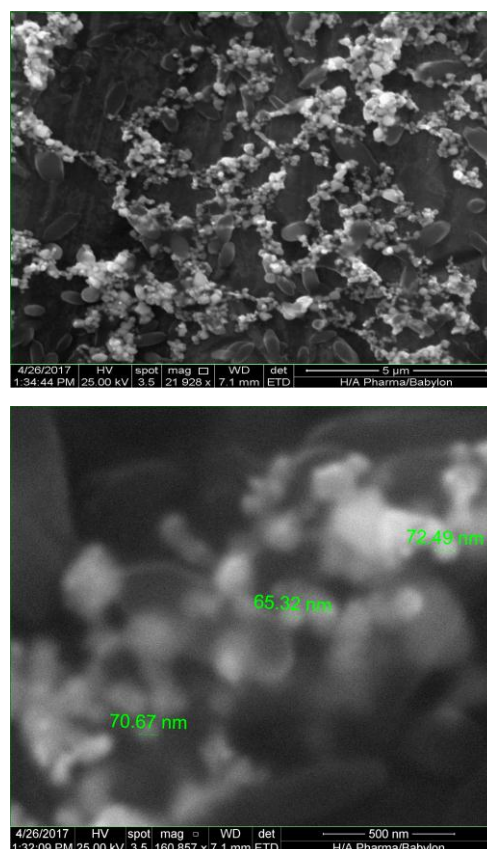


Figure 2: SEM image of Ag nanoparticles.

### Antimicrobial activity of silver nanoparticles

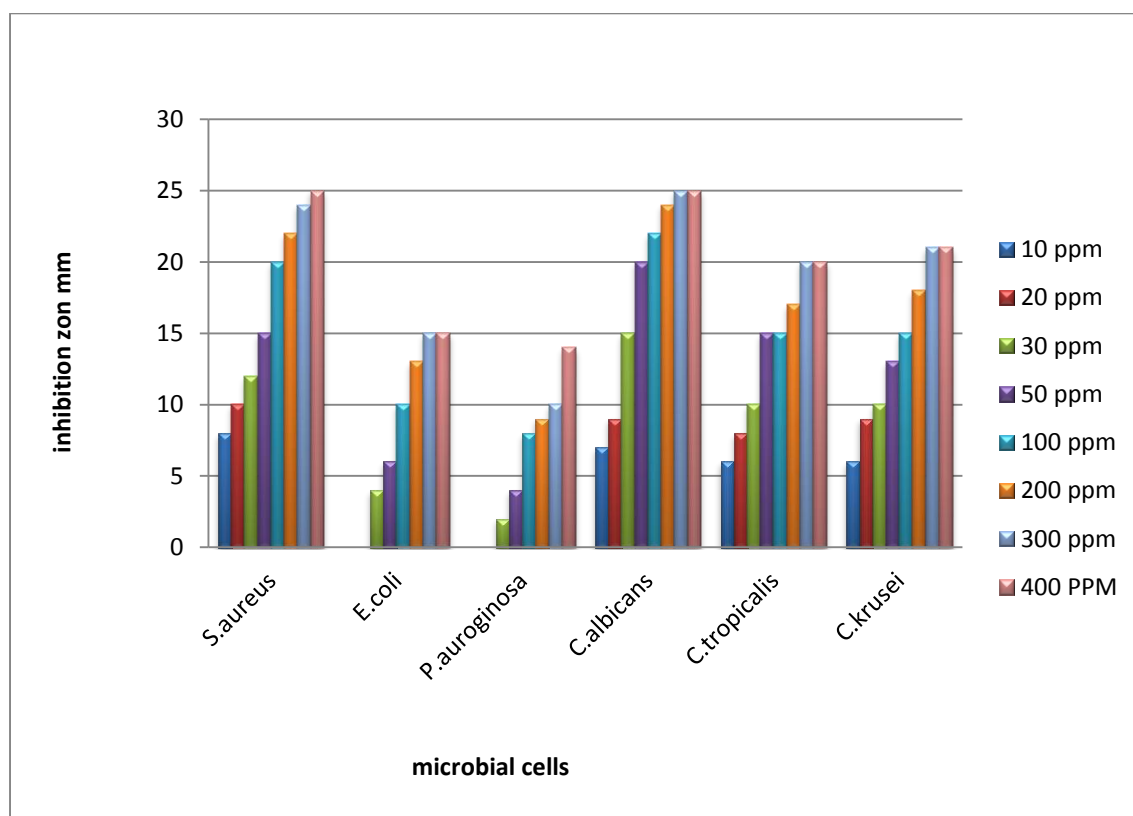
Table (1) showed the antimicrobial properties of Ag- NPs were studied against isolates, *S. aureus* ( MRSA), *E.coli* , *P.aeruginosa* *Candida albicans*, *Candida tropicalis* , *Candida krusei*. The Minimum inhibitory concentration of Ag –NPs against *Candida albicans*, *Candida tropicalis* , *Candida krusei* were found 30 ppm , 30ppm , 30 ppm , respectively. The MICs of AgNPs were 100 ppm for other bacterial test strains, including *S.aureus* (MRSA) , *E.coli* , *Pseudomonas aeruginosa* .While the Minimum Bacterial / Fungicidal Count of Ag –NPs against *Candida sp.* were found 300 ppm . It was observed from MIC values obtained in broth dilution method that antimicrobial efficiency of Nanosilver solution was higher against yeasts compared to bacteria . Fig (3) showed inhibition zones increases with increasing concentration of silver. Similar results have been reported on some bacteria and fungi ( 28- 30 ).Some authors demonstrate different values in MIC results for determining antimicrobial activity of silver nanoparticles against bacteria and fungi due to general physiological differences in the cell wall membrane of microorganisms( 31). Indeed to that other scientific studies referred to Ag NP activity depend on the their concentration , size and shape (32- 35 )

Also antimicrobial activity of silver ions bind to the protein and nucleic acid negatively charged ,causing structural

changes and deformations in the (wall, membranes , nucleic acids) . Silver ions interact with a number of electron donor functional groups such as hydroxyls, imidazoles, thiols, phosphates and indoles. Also AgNPs induce and release of reactive oxygen species (ROS), forming free radicals . AgNPs can enter the bacteria or fungi and damage of its cellular structures, inhibition of protein synthesis as a result of ribosomes denaturation, as well as translation and transcription will be blocked by the binding with genetic material of the bacteria(36- 39) .

**Table 1:** Determination of MIC and MBC for silver nanoparticles for isolates

Isolate	MIC Conc.(ppm)	MBC Conc.(ppm)
<i>C. albicans</i>	30	300
<i>C. tropicalis</i>	30	300
<i>Candida krusei</i>	30	300
<i>S. aureus</i>	100	300
<i>E.coli</i>	100	300
<i>P. auroginosa</i>	100	300



**Figure 3:** Antimicrobial activity of Ag- NPs against bacteria and fungi

## CONCLUSION

In this study, simple and effective method of synthesis of AgNPs solution using a chemical reduction method, was tested against six isolates. Different degrees of sensitivity were shown among isolates. The study concludes that Ag-NPs have potent antimicrobial activities against tested isolates. Therefore, it was proposed as an alternative drug for antimicrobial activity and can inhibit the Gram positive, gram negative growth and yeast growth.

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